

**THERMODYNAMIC AND ELECTROPHYSICAL CHARACTERISTICS  
OF THE Nd<sub>2</sub>O<sub>3</sub>–MgO–NiO–Mn<sub>2</sub>O<sub>3</sub> SYSTEM**

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Modification and combination of alloying elements in NdMnO<sub>3</sub>-based compounds allow for the production of materials with valuable physical and chemical properties.

In the Nd<sub>2</sub>O<sub>3</sub>-MgO-NiO-Mn<sub>2</sub>O<sub>3</sub> system, the compound NdMg<sub>2</sub>NiMnO<sub>6</sub> was synthesized using a high-temperature ceramic method at 400–1200 °C.

The temperature dependence of the heat capacity of NdMg<sub>2</sub>NiMnO<sub>6</sub> in the range 298.15–673 K was studied using the IT-C-400 calorimeter, the principle of operation of which is described in detail in [1].

The temperature dependence of the heat capacity of NdMg<sub>2</sub>NiMnO<sub>6</sub> shows λ-shaped peaks at 348 and 473 K, which are probably related to phase transitions. Using these peaks, the following equations for  $C_p^0 \sim f(T)$ , [J/(mol·K)], have been calculated:

$$C_p^0 = -(75 \pm 3) + (1024 \pm 41) \cdot 10^{-3} T, \quad (298-348 \text{ K})$$

$$C_p^0 = (1004 \pm 40) - (2077 \pm 84) \cdot 10^{-3} T, \quad (348-398 \text{ K})$$

$$C_p^0 = (2109 \pm 85) - (2286 \pm 92) \cdot 10^{-3} T - (1618 \pm 65) \cdot 10^5 T^{-2}, \quad (398-473 \text{ K})$$

$$C_p^0 = (1187 \pm 48) - (1867 \pm 75) \cdot 10^{-3} T, \quad (473-548 \text{ K})$$

$$C_p^0 = (541 \pm 22) + (33 \pm 1,32) \cdot 10^{-3} T - (1185 \pm 48) \cdot 10^5 T^{-2}. \quad (548-673 \text{ K})$$

$S^0(298,15)$  of the NdMg<sub>2</sub>NiMnO<sub>6</sub> compound is estimated by the method of ion entropy increments of Kumok. Using the experimental dependence of  $C_p^0 \sim f(T)$  and the calculated  $S^0(298,15)$ ,  $C_p^0(T)$ ,  $S^0(T)$ ,  $H^0(T) - H^0(298,15)$ ,  $\Phi^{xx}(T)$  are determined.

Using the LCR-781 setup in the 293–483 K (1, 5, and 10 kHz) range, it was found that the dielectric constant of NdMg<sub>2</sub>NiMnO<sub>6</sub> increases sharply with increasing temperature ( $\epsilon \approx 1.16 \times 10^7$  at 483 K), while the resistance decreases ( $1.53 \times 10^6 \rightarrow 1.38 \times 10^4$  Ohm), confirming the semiconductor nature of the conductivity.

1. Kasenova Sh.B., Sagintaeva Zh.I., Kasenov B.K., Kuanyshbekov E.E., Ordabaeva A.T., Isabaeva M.A. Heat Capacity and Thermodynamic Characteristics of Sodium and Potassium Nickelite-Manganites of Neodymium of NdNa<sub>2</sub>NiMnO<sub>5</sub> and NdK<sub>2</sub>NiMnO<sub>5</sub> // Applied Sciences, 2025, 15, 6751. <https://doi.org/10.3390/app15126751>

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